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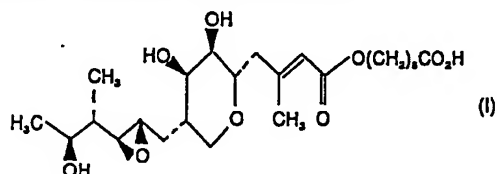
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**(54) Silver pseudomunate, compositions containing it and its use in treating pseudomonal infections.**

57 Pseudomonic acid (I) is an antibiotic produced by aerobically culturing *Pseudomonas fluorescens*.



A process is provided for producing silver pseudomonate which process comprises reacting silver ions and pseudomonic acid or pseudomonate ions in aqueous solution and thereafter recovering the silver pseudomonate so formed.

Also provided is a method for treating wounds or burns infected with *Pseudomonas* organisms comprising administering a non-toxic anti-pseudomonally effective amount of silver pseudomonate to the wound or burn.

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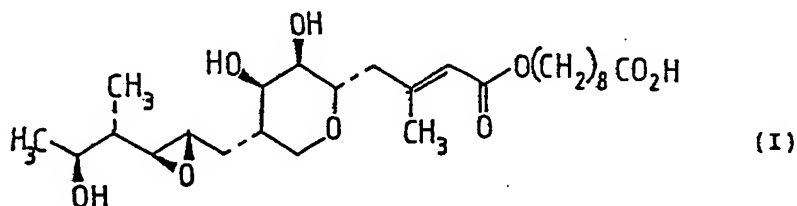
## TITLE MODIFIED

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COMPOUND AND USE

The present invention relates to silver pseudomonate, compositions containing it and its use in treating pseudomonal infections.

Pseudomonic acid is an antibiotic produced by aerobically culturing Pseudomonas fluorescens. The compound, of formula (I) below, and its salts and esters are disclosed and claimed in UK Patent No. 1 395 907.



Whilst pseudomonic acid and its salts and esters are active against a variety of human and animal pathogens (see for instance UK Patent Nos. 1 577 730 and 1 577 545), they are not active at useful levels against Pseudomonas species.

Pseudomonas organisms tend to infect burns and wounds. Such infections are often difficult to treat as the organisms are not particularly susceptible to antibiotics.

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02           It has now surprisingly been found that silver  
03 pseudomonate is active against Pseudomonas organisms,  
04 especially Pseudomonas aeruginosa, the causative agent  
05 of 'blue pus' infections.

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07           The silver salt of pseudomonic acid has not been  
08 specifically disclosed in the above patents or any  
09 other publications and is, therefore, novel.

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11           Accordingly the present invention provides, in one  
12 aspect, silver pseudomonate.

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14           The invention also provides silver pseudomonate  
15 for use in the treatment of the human or animal body.

16

17           Apart from its surprising activity against  
18 Pseudomonas, silver pseudomonate has a similar spectrum  
19 of activity against pathogens to those of pseudomonic  
20 acid and sodium pseudomonate.

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22           Accordingly the present invention also provides  
23 silver pseudomonate for use in treating the human or  
24 animal body, especially for treating infected wounds  
25 and burns.

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27           The invention also provides a process for  
28 producing silver pseudomonate which process comprises  
29 reacting silver ions and pseudomonic acid or  
30 pseudomonate ions in aqueous solution and thereafter  
31 recovering the silver pseudomonate so formed.

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33           Suitably the process is effected by adding a  
34 source of silver ions to an aqueous solution of  
35 pseudomonic acid or a pseudomonate salt, especially  
36 sodium pseudomonate.

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02           Suitably the solution of pseudomonic acid or  
03 pseudomonte ions is the product of aerobically  
04 culturing Pseudomonas fluorescens (NCIB 10586). Such a  
05 solution may be the culture medium in which the  
06 organisms have been grown or it may have been produced  
07 by purifying such a medium for instance by extracting  
08 pseudomonic acid from such a culture medium using a  
09 polar, organic, water-immiscible solvent as described  
10 in EP 0 005 614. Alternatively the solution of  
11 pseudomonic acid or pseudomonte ions may be produced  
12 by dissolving pseudomonic acid or preferably a salt  
13 thereof, in an aqueous solvent. Preferably the  
14 solution is produced by dissolving pure sodium  
15 pseudomonte in water.

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17           The source of silver ions is preferably a soluble  
18 silver salt such as silver nitrate or silver carbonate.

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20           The invention further provides silver pseudomonte  
21 in substantially pure form, preferably at least 75%  
22 pure, more preferably at least 90% pure, most  
23 preferably at least 95% pure.

24

25           If precipitated from solution containing solvents  
26 other than water, the silver pseudomonte may be  
27 produced in a solvated form including a hydrated form.  
28 If precipitated from aqueous solution the silver  
29 pseudomonte may be in a hydrated form.

30

31           Accordingly the invention further provides  
32 solvated, including hydrated, silver pseudomonte.

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34           Silver pseudomonte may be administered as the  
35 pure compound (hereinafter referred to as the 'drug')  
36 or it may be administered as a pharmaceutical  
37 composition in association with a suitable carrier.

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Accordingly the invention also provides a pharmaceutical formulation comprising silver pseudomonate and a pharmaceutically acceptable carrier therefor.

As used herein the term 'pharmaceutically acceptable' includes 'veterinarily acceptable'.

The formulations may be adapted for administration by any route, and would depend on the disease being treated. Normally, the formulations will be presented as topical solutions or suspensions for application to the skin, ears or eyes. Alternatively the formulations may be dry powders for application as an aerosol, or they may be presented as impregnated dressings for wounds and burns.

For topical application to the skin the drug may be made up into a cream, lotion or ointment. Cream or ointment formulations that may be used for the drug are conventional formulations well known in the art, for example, as described in standard text books of pharmaceuticals and cosmetics, such as Harry's Cosmeticology published by Leonard Hill Books, and the British Pharmacopoeia. Alternatively the drug may be applied as a dry powder from an aerosol using conventional diluents and propellants.

For topical application to the ear, the drug may be made up into a solution or suspension in a suitable liquid carrier, such as water, glycerol, diluted ethanol, propylene glycol, polyethylene glycol or fixed oils.

For topical application to the eye, the drug is formulated as a solution or suspension in a suitable, sterile aqueous or non-aqueous vehicle. Additives, for

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02 instance buffers such as sodium metabisulphite or  
03 disodium edetate; preservatives including bactericidal  
04 and fungicidal agents, such as phenylmercuric acetate  
05 or nitrate or chlorhexidine, and thickening agents such  
06 as hypromellose may also be included.

07

08 Particularly suitable topical formulations  
09 comprise silver pseudomonate and at least 1% by weight  
10 of a poly (substituted or unsubstituted alkylene)  
11 glycol or a derivative thereof.

12

13 As used herein the term 'poly (substituted or  
14 unsubstituted alkylene) glycol' refers to polymers  
15 having the following repeating unit

16

17  $-(CH_2)_nO-$

18

19 wherein n is an integer, preferably 2 or 3 and to such  
20 polymers wherein one or more methylene groups of each  
21 repeating unit is substituted. Suitable substituents  
22 include alkoxy groups such as methoxy as in  
23 polymethoxypropylene glycol. Such polymers are known  
24 by a variety of names, for instance when  $n = 2$ , as  
25 polyethylene glycol, polyoxyethylene, polyoxyethylene  
26 glycol and macrogol and, when  $n = 3$ , as polypropylene  
27 glycol, polyoxypropylene and polyoxypropylene glycol.  
28 All these are useful in the invention as are  
29 derivatives of these polymers.

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31 Suitable derivatives include ethers and esters of  
32 the poly (substituted or unsubstituted alkylene)  
33 glycols, such as the macrogol ethers and esters, for

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02 instance cetomacrogol, glycofurool, the 'Tweens'\* and  
03 block copolymers including poly (substituted or  
04 unsubstituted alkylene) glycols such as Poloxamers  
05 which are block copolymers of polyethylene glycol and  
06 polypropylene glycol for instance the 'Pluronics'\*, and  
07 cross-linked polyethylene glycol.

08

09 The poly (substituted or unsubstituted alkylene)  
10 glycols and derivatives thereof may be used singly or  
11 various grades and types may be used in combination to  
12 achieve the desired physical properties of the  
13 formulation.

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15 Preferably the formulation comprises polyethylene  
16 glycol or a derivative thereof.

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18 Suitably the formulation comprises from 0.01 to  
19 50% by weight of silver pseudomonate, preferably 0.1 to  
20 25%, more preferably 0.5 to 10% and most preferably  
21 about 2% by weight of silver pseudomonate calculated as  
22 the free acid. Such formulations comprising only  
23 silver pseudomonate and a poly (substituted or  
24 unsubstituted alkylene) glycol or derivative thereof  
25 will, of course, contain up to 99.99% of the poly  
26 (substituted or unsubstituted alkylene) glycol or  
27 derivative thereof.

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29 The formulation may comprise additional  
30 therapeutic agents such as antibacterial, antifungal,  
31 antiviral and antiinflammatory agents, for instance  
32 chlortetracycline, miconazole, idoxuridine and  
33 phenazone, provided that these are compatible with the

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35 \* 'Tween' and 'Pluronic' are trade names for the above  
36 types of polymer.

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silver pseudomonte. Silver Pseudomonte tends to undergo a rearrangement reaction in the presence of acid and accordingly acidic agents are unlikely to be compatible with silver pseudomonte.

In a particular aspect the invention provides a topical formulation as described above wherein silver pseudomonte is the sole therapeutic agent.

In another aspect the invention provides a topical formulation comprising silver pseudomonte and at least 1% by weight of polyethylene glycol or a derivative thereof.

Polyethylene glycols (PEG's) and derivatives thereof are commercially available in a variety of chain lengths and with a variety of consistencies, for instance:-

Polyethylene Glycols:-

Liquids	Semisolids	Hard Solids
PEG 200 PEG 300 PEG 400	PEG 1000 PEG 1540	PEG 4000* PEG 6000

Polyethylene Glycol derivatives:-

Derivative	Chemical Composition	Consistency
Glycofurol	Tetrahydrofurfuryl alcohol polyethylene glycol ether	Liquid
Tween 60	Polyoxyethylene Sorbitan monostearate	Semi-solid
Tween 80	Polyoxyethylene Sorbitan monooleate	Liquid

\* PEG 4000 is the B.P. nomenclature for PEG with mean molecular weight of 3350. This material is also known as PEG 3350 in U.S.P. nomenclature.



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These may be used singly or admixed in suitable proportions to achieve the desired consistency of formulation.

The formulations of the present invention may contain appropriate conventional additives such as preservatives, solvents to assist drug penetration and emollients in ointments and creams. The formulations may also contain compatible conventional carriers, such as cream or ointment bases and ethanol or oleyl alcohol for lotions. Such carriers may be present as from about 1% up to about 98% of the formulation. More usually they will form up to about 80% of the formulation.

Particularly suitable formulations according to the present invention comprise at least 1% by weight of PEG or a mixture of PEG's, from 0 to 25% by weight of a PEG derivative or mixture of PEG derivatives and from 0.5 to 10% by weight of silver pseudomonate calculated as the free acid.

Preferably the silver pseudomonate represents 1 to 5% of the formulation, most preferably about 2% of the formulation calculated as the free acid.

Formulations of the invention may be produced by conventional pharmaceutical techniques. Thus ointments and creams are conveniently prepared by melting and mixing together the solid or semi-solid PEG's or PEG analogues or derivatives, and stirring in the therapeutic agent and any other ingredients. The product is then slowly cooled and filled into containers such as collapsible metal or plastic tubes.

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Liquid preparations, such as ear and eye drops, are produced by dissolving the therapeutic agent in the liquid PEG's or PEG analogues or derivatives and the other ingredients are then added. The resulting solution or suspension is distributed into glass or plastic bottles or in single dose packs such as soft gelatin capsules which are then heat sealed.

If necessary the formulation may be milled at any suitable stage of the process.

A suitable sterilisation procedure may be included in the above processes if necessary. Alternatively raw materials are obtained in sterile conditions and the formulations are produced aseptically.

The dosage employed for formulations administered topically will, of course, depend on the size of the area being treated. For the ears and eyes each dose will typically be in the range from 10 to 100 mg of the drug.

The present invention further provides a process for producing a pharmaceutical formulation which process comprises bringing into association silver pseudomonate and a pharmaceutically acceptable carrier therefor.

The present invention also provides a method for treating pseudomonal infections of human or non-human animals comprising administering a non-toxic anti-pseudomonally effective amount of silver pseudomonate to an infected human or non-human animal.

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02           In a particular aspect the invention provides a  
03           method for treating wounds or burns infected with  
04           Pseudomonas organisms comprising administering a  
05           non-toxic anti-pseudomonally effective amount of silver  
06           pseudomonate to the wound or burn.

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08           Preferably the above methods are effected by  
09           applying a topical formulation to the infected area.

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11           The invention will now be illustrated with  
12           reference to the following Examples and Biological  
13           data.

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Example 1

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Silver Pseudomonate A

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Sodium pseudomonate A (1.82g, 4 mmol) and silver nitrate (0.68g, 4 mmol) were stirred in distilled water for 30 min resulting in the formulation of a white gelatinous precipitate. The mixture was centrifuged, the aqueous layer removed and the residue washed with distilled water. The suspension was centrifuged and the residual solid was dried over phosphorus pentoxide under high vacuum for 2 days to yield silver pseudomonate A, m.p. 164-166°C, (855 mg, 35%);  $\nu_{\max}(\text{KBr})$  3400, 1710, 1645, 1515  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}(\text{CD}_3)_2\text{SO}$  5.68 (1H, s, H2), 2.12 (3H, s,  $\text{CH}_3$ -15), 1.1 (3H, d,  $\text{CH}_3$ -14), 0.85 (3H, d,  $\text{CH}_3$ -17) (Found: C, 49.6; H, 6.7; Ag, 17.8.  $\text{C}_{26}\text{H}_{43}\text{O}_9\text{Ag}$  requires C, 51.4; H, 7.1; Ag, 17.8%).

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Example 2

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Liquid Formulation

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Example 3

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Ointment Formulation

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% w/w

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PEG 400

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PEG 4000

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Silver pseudomonate

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The formulation may be produced by melting the mixture of PEG's and stirring in the silver pseudomonate.

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Example 4

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Lotion Formulation

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% w/w

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PEG 400

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Ethanol

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Silver pseudomonate

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Example 5

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Drop Formulation

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% w/w

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PEG 400

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Glycofurol

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Silver pseudomonate

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Example 6

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% w/w

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Cetomacrogol emulsifying ointment

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Polyethylene glycol 200

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Silver pseudomonate

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BIOLOGICAL DATA

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- a) The minimum inhibitory concentrations (MICs) of silver pseudomonate and sodium pseudomonate were determined against 20 strains of Pseudomonas aeruginosa in Blood Agar Base. Typical results are presented in Table 1. Silver pseudomonate was more active than sodium pseudomonate against all strains tested.
- b) MIC's of silver and sodium pseudomonate against various pathogenic bacteria were determined by standard methods. Typical results are presented in Table 2.

Table 1

The activity of Sodium Pseudomunate and Silver  
Pseudomunate against 20 strains of  
Pseudomonas aeruginosa:

Typical MIC's

Pseudomonas aeruginosa	MIC* ug/ml	
	Sodium Salt	Silver Salt
NCTC 10662	12,800	128
Dalglish	>128	128
PU7	>128	128
W985	>128	128
S41	>128	128
R60	>128	128
Pu4	>128	128
R59	>128	64
T3	>128	128
R3	6,400	128
R139	>128	128
R22	>128	128
W995	>128	128
59	>128	128
125	>128	128
4	>128	128
Fr13	6,400	128
D25	>128	128
ATCC 27853	>128	128
W996	>128	128

\* MIC determined in serial dilution in Blood Agar  
Base. Inoculum of 0.001 ml of an overnight Tryptone  
Soya Broth Culture. Incubated at 37°C overnight.



Table 2Typical MIC's (µg/ml) against Human Bacteria

Organism	<u>Pseudomonate Salt, MIC (µg/ml)</u>	
	Silver	Sodium
E. coli NCTC 10418	128	125
P. mirabilis 889	128	125
K. aerogenes A	128	250
Ps. aeruginosa NCTC 10662	128	12800
Pasteurella multocida 1633	0.5	0.25
Haemophilus influenzae Wy21	0.12	0.12
Bacillus subtilis 6633	0.25	0.25
Corynebacterium xerosis 9755	128	>125
Staph. aureus Oxford	0.5	0.25
Staph. aureus Russell	0.5	0.25
Staph. aureus W2827	0.5	0.25
Strep. faecalis I	64	50
Strep. pyogenes R80/421-A	0.25	0.25
Strep. agalactiae 2788-B	1.0	0.5
Strep. spp. 64/848-C	1.0	0.5